

ARCHIVE 2006 Tutorial

"Fundamental Properties Of Electrical Contacts"

Dr. Roland Timsit
President

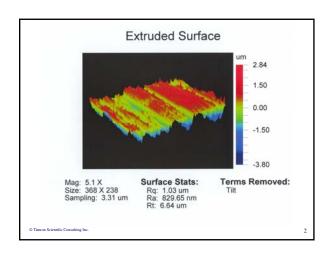
Timron Scientific Consulting Inc.

COPYRIGHT NOTICE

- The papers in this publication comprise the proceedings of the 2006 BiTS Workshop. They reflect the authors' opinions and are reproduced as presented, without change. Their inclusion in this publication does not constitute an endorsement by the BiTS Workshop, the sponsors, BiTS Workshop LLC, or the authors.
- There is NO copyright protection claimed by this publication or the authors. However, each presentation is the work of the authors and their respective companies: as such, it is strongly suggested that any use reflect proper acknowledgement to the appropriate source. Any questions regarding the use of any materials presented should be directed to the author/s or their companies.
- The BiTS logo and 'Burn-in & Test Socket Workshop' are trademarks of BiTS Workshop LLC.

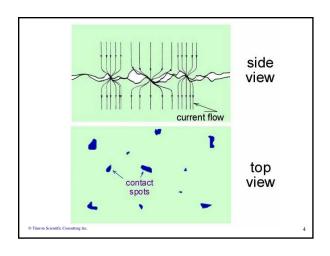


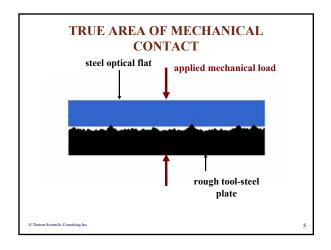
Fundamental Properties of Electrical Contacts R.S. Timsit Timron Scientific Consulting Inc. Toronto, ON, CANADA

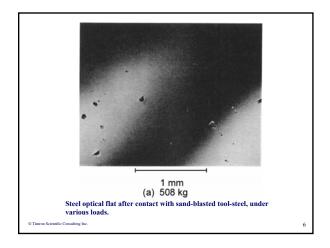


Because all solid surfaces are rough on the microscale, two mating solid surfaces make contact only where the peaks of small surface asperities (roughness) touch one another.

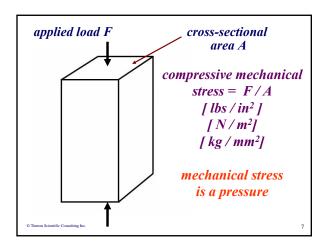
© Timron Scientific Consulting In

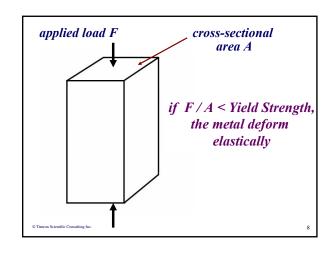


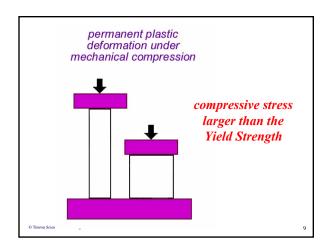


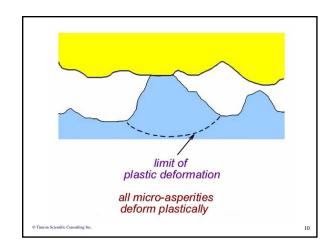


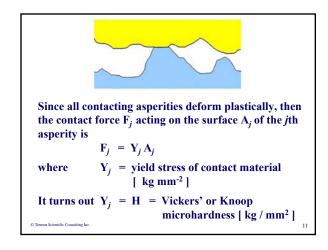


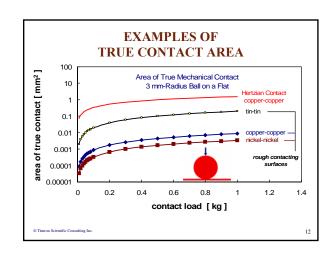




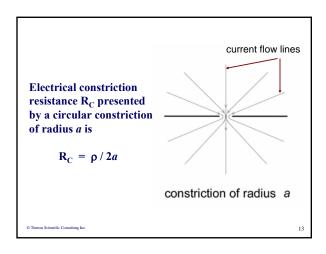


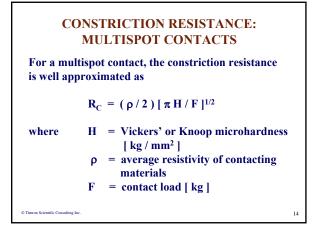


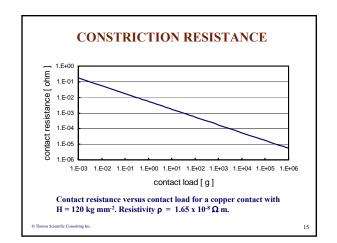


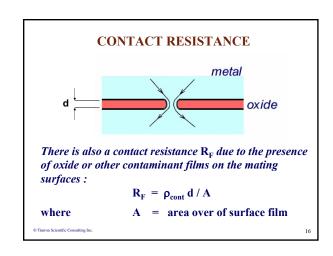


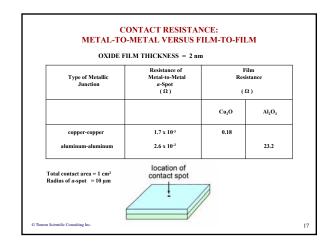


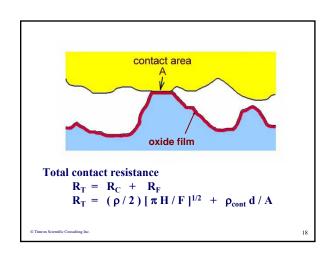






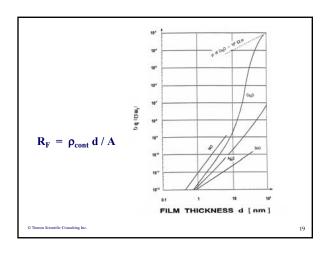


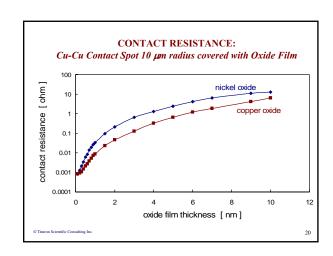




2006

Tutorial I





SURFACE OXIDE FILMS

To enhance electrical contact reliability

- <u>do not tolerate</u> surface contaminant films i.e. do not expect conduction though them
- abrade/remove all surface films, in particular oxide layers

© Timron Scientific Consulting Inc.

GROWTH OF INTERMETALLIC COMPOUNDS The width X of an intermetallic

The width X of an intermetallic layer or an interdiffusion band increases with time t as

 $X^2 = kt$ with k = interdiffusion constant= $k_0 \exp(-Q/RT)$

> Q = activation energy R = gas constant T = absolute temperature

(a) after 765 hours at 250°C

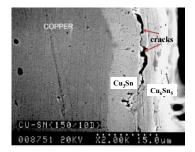
BRASS

D
C
C
BRASS

D
C
C
A
AI

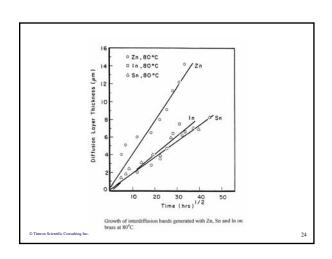
(b) after 6 hours at 400°C

INTERMETALLICS FORMATION

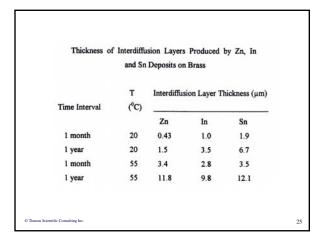


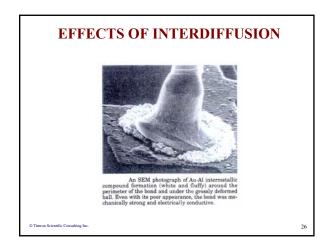
Intermetallics growth at a copper/tin interface.

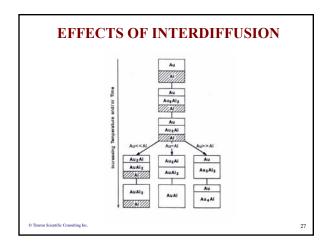
Timron Scientific Consulting I

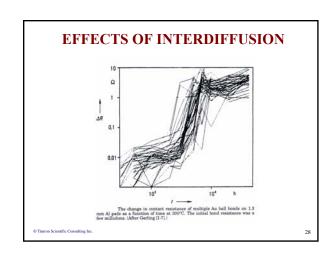




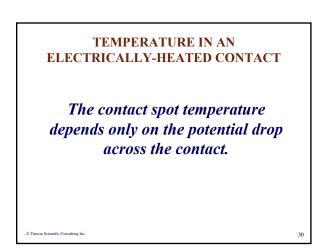


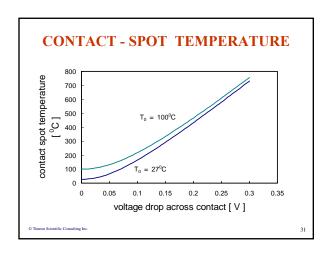


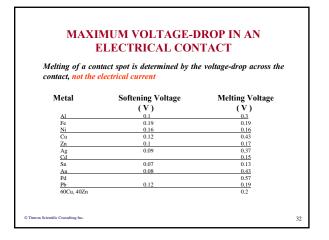




TEMPERATURE IN AN ELECTRICALLY-HEATED CONTACT a contact spot may be considered a "thermally-insulated" resistor of value R 1. if the resistor is made of copper and R = 10 x 10° d ohm, it is found that the resistor melts at a current of 43,000 A 2. if the resistor is made of copper and R = 1 ohm, it is found that the resistor melts at a current of 0.43 A What is the common factor describing melting of the contact spot?







TEMPERATURE IN AN ELECTRICALLY-HEATED CONTACT

thermal risetime τ of a contact spot of radius "a"

 $\tau = C a^2 / 4\lambda$

where C = conductor heat capacity

 λ = thermal conductivity

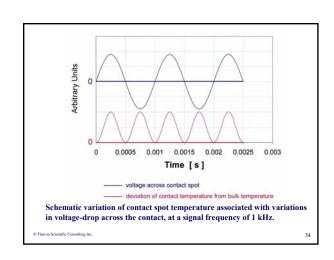
For copper, $C = 3.44 \text{ J cm}^{-3} {}^{0}C^{-1}$

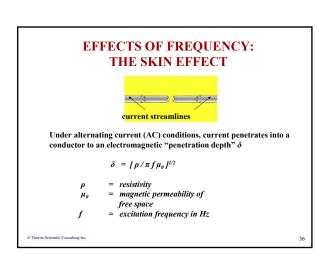
 $\lambda = 4 \text{ W cm}^{-1} \, {}^{0}\text{C}^{-1}$

so that $\tau = 2.2 \times 10^{-7} \text{ s for a contact spot}$

with $a = 10 \mu m$

© Timron Scientific Consulting Inc.

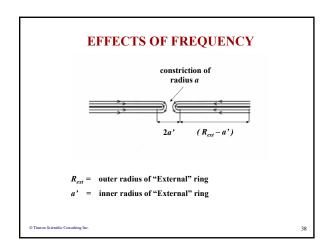






EFFECTS OF FREQUENCY: THE SKIN EFFECT

Variation of Skin Depth with Frequency for a Metal of Resistivity 3 x $10^{-8} \Omega$ m		
f	Skin Depth δ	
[Hz]	[µm]	
60	11254	
10 ³	2757	
104	872	
105	276	
10^{6}	87	
107	28	
108	8.7	
109	2.8	



EFFECTS OF FREQUENCY

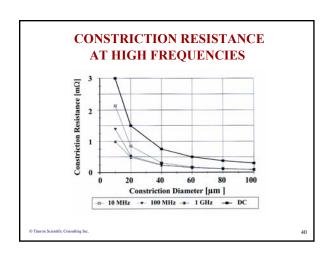
Total Connection Resistance = Constriction Resistance + Resistance of "External"Ring

Resistance of "External" Ring = $(\rho / 2\pi \delta) \ln (R_{ext}/a')$

 $\begin{array}{lll} \delta & = & \text{electromagnetic penetration depth} \\ R_{ext} & = & \text{outer radius of "External" ring} \\ a' & = & \text{inner radius of "External" ring} \end{array}$

 ρ = resistivity

© Timron Scientific Consulting Inc.

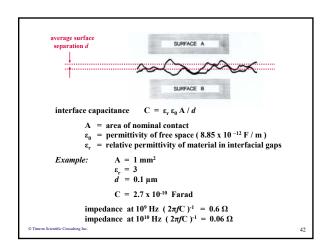


CONSTRICTION RESISTANCE VS. CONNECTION RESISTANCE AT HIGH FREQUENCIES

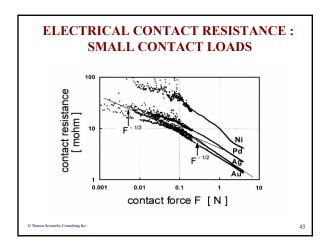
Signal Frequency f [Hz]	Constriction Resistance [mΩ]	Connection Resistance [mΩ]
107	2.2	3.2
108	1.4	4.7
109	1.0	11.2

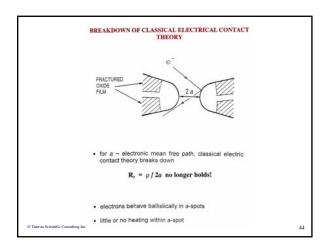
 $\rho = 3 \times 10^{-8} \Omega \text{m}$ $a = 5 \mu \text{m}$ $R_{ext} = 100 \mu \text{m}$.

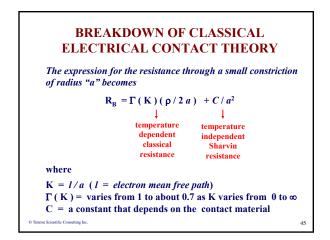
D Timmon Scientific Conculting In

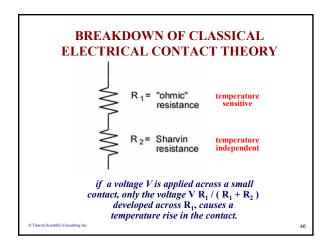


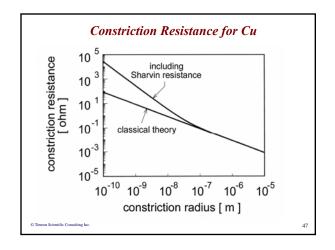
41

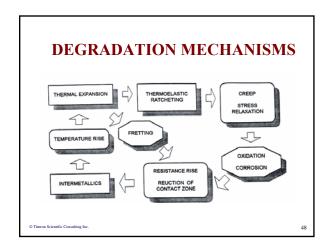










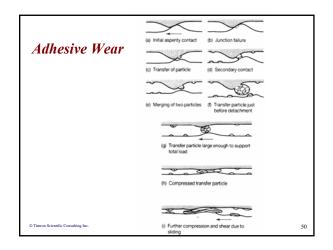


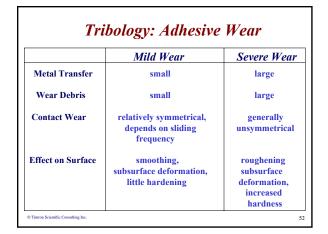
Tribology

Most common types of wear in metal sliding:

- I. Abrasive Wear relevant only to high power connectors
- II. Adhesive Wear
- III. Fretting Wear
- IV. Erosion generally not relevant to connectors
- V. Lubricated Wear not relevant to connectors

© Timron Scientific Consulting In

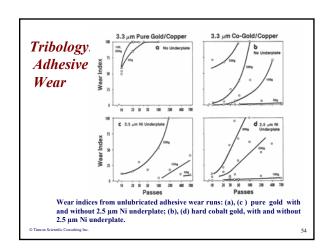




Tribology: Adhesive Wear Transition Loads

	Pure Gold	Hard Gold
Clean Surface	5 g	10 g
Contaminated	10 – 50 g	25 – 300 g
Lubricated	100 – 500 g	500 – 2000 g

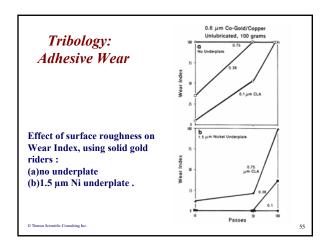
O Timeon Scientific Consulting Inc

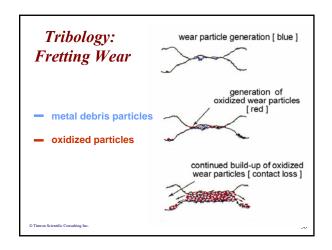




2006

Tutorial I





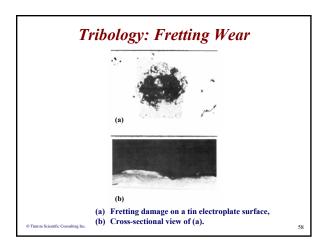
Tribology: Fretting Wear

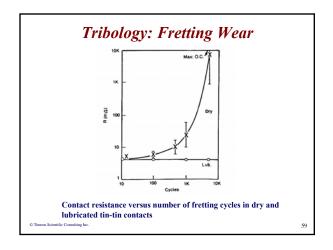
Fretting Wear is generated by small-amplitude movement leading to the formation of small debris particles at a mechanical interface. In electronic connectors, the amplitude of this micromotion ranges from a few μ m to about 100 μ m.

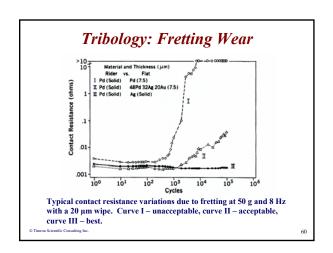
Micromotion is caused by external vibrations or by changing temperature due to differences in thermal expansion coefficients of the mating materials.

Oxidation of fretting debris leads to increased electrical contact resistance.

© Timron Scientific Consulting Inc.





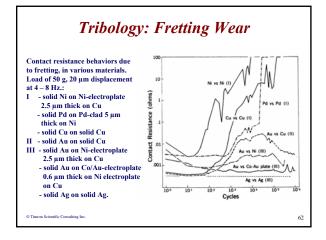


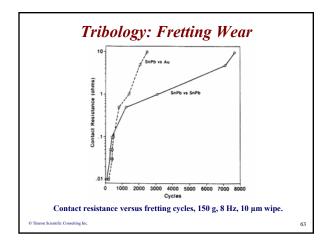
Tribology: Fretting Wear Silver

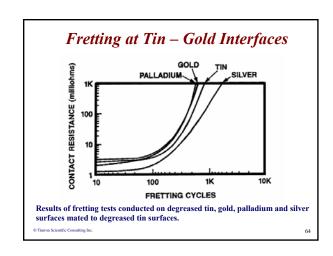
Silver is the most stable material in fretting, since it is relatively wear resistant, does not oxidize readily, and does not form frictional polymers. It displays excellent behaviour when mated to itself. Silver is prone to tarnish in the presence of even minute amounts of sulfur and chlorine compounds. This limits the use of silver in electronic connectors.

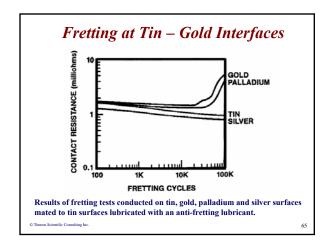
Silver is widely used as a finish on aluminum busbar contacts.

© Timron Scientific Consulting In



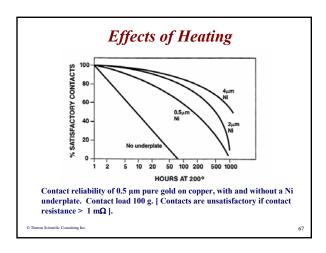


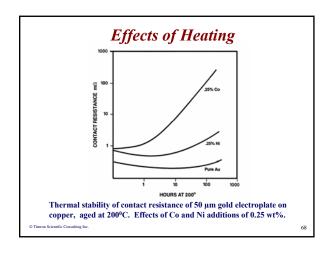




Tribology: Fretting Wear Gold Gold approaches silver in its stability. Although it is known that has organic materials, aerosols and other contaminant layers accumulate on gold surfaces to increase contact resistance, these contaminants are usually eliminated by rubbing. It has been claimed that traces of polymer form when gold contacts are rubbed together in benzene vapor or immersed in an oil. No deleterious effect of this polymer on contact resistance has been detected.



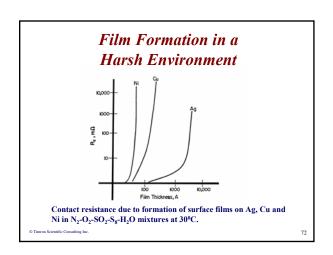




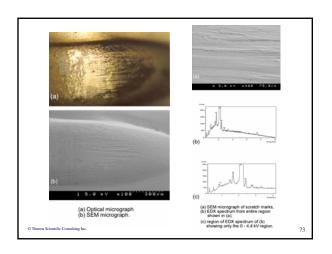
OXIDATION / CONTAMINATION IN AIR THICKNESS (nm) copper:- oxide forms immediately - thickness depends on T⁰C $10^3 h$ 10⁵ h temperature 20 2.2 55 3.5 17 85 **8.7** 69 100 15.0 130 tin: - oxide growth is initially slow 20 4.2 6.1 - depends on temperature 55 10.3 14.6 Sn 85 18.8 26.0 100 25.0 36.0 © Timron Scientific Consulting Inc.

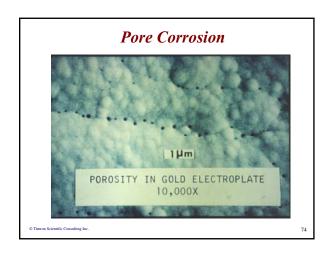
OXIDATION / CONT	TAMINATIO	ON IN	AIR
		THICK	NESS (nm)
	T ⁰ C	10 ³ h	10 ⁵ h
<u>nickel</u> :- oxide growth is self-limiting - weak dependence on temperature	20 Ni 55 85 100	1.6 2.1 2.7 3.4	15.0 21.0 27.0 34.0
<u>silver</u> :- Ag ₂ S formation - formation of Ag ₂ O in	presence of ozo	one	
© Timron Scientific Consulting Inc.			70

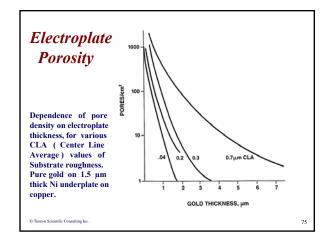
<u>Material</u>	Reactivity to Environmental Pollutants
Ag, Pd-Ag, Au-Ag, Au-Cu	S, H_2S, Cl_2
Pd, Pd-Cu	NO_2 , SO_2 , Cl_2
Cu	S , SO_2 , H_2S , Cl_2
Ni	SO_{2} NO_{2} Cl_{2}
Sn, Sn-Pb	NO ₂ , SO ₂ , Cl ₂ , H ₂ O

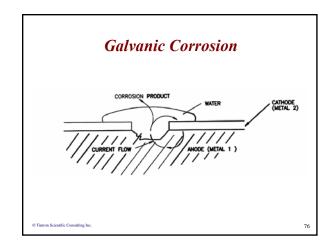


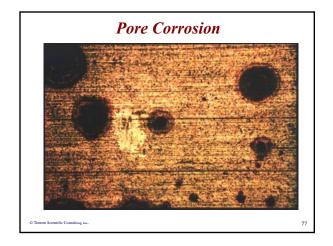


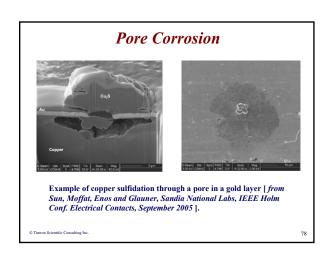












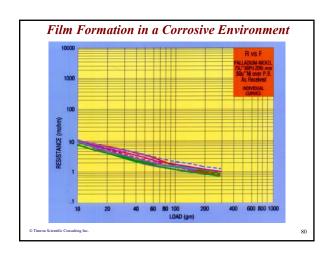


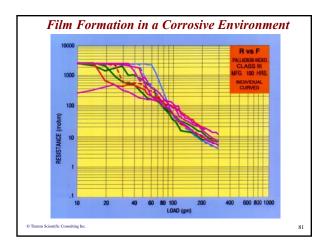
Connector Testing: Mixed Flow Gas (MFG) Composition MFG TEST CONDITIONS FOR ACCELERATING THE EFFECTS OF ENVIRONMENTAL CLASSES II, III AND IV

	Severity Class		
	11	III	IV
Temperature (°C)	30±2	30±2	40±2
Relative Humidity (%)	70±2	75±2	75±2
Chlorine (Cl ₂), ppb	10±3	20±5	30±5
Nitrogen Dioxide (NO ₂), ppb	200 ± 50	200 ± 50	200 ± 50
Hydrogen Sulfide (H ₂ S), ppb	10±5	100 ± 20	200 ± 20

ppb=parts per billion of each gas in air.

© Timron Scientific Consulting Ir





SIGNIFICANCE OF BURN-IN IN ELECTRICAL CONTACTS

Selected Burn-In Methods

- Pass a large current through contacts possible beneficial effects:
 - slight overheating of contact spots, causing negligible metallurgical effect, may soften contact spots and increase the true contact area to reduce contact resistance
 - slight differential expansion in contact region may cause local abrasion of surface contaminant films and reduce contact resistance

© Timron Scientific Consulting In

82

SIGNIFICANCE OF BURN-IN IN ELECTRICAL CONTACTS

Selected Burn-In Methods

- Pass a large current through contacts possible deleterious effects:
 - overheating of contact spots with possible metallurgical changes in the contact region
 - increased oxidation
 - overheating of contact springs or connector components with possible decrease in contact force due to stress relaxation or metal-creep

Timron Scientific Consulting Inc

• passing a large burn-in current through some types of contacts, such as those using low meltingpoint materials, may be particularly deleterious to contact reliability



SIGNIFICANCE OF BURN-IN IN ELECTRICAL CONTACTS

Selected Burn-In Methods

- Reciprocating motion of pin in socket while passing current, but without contact disconnect major beneficial effect:
 - disperse surface contaminant films and reduce contact resistance

© Timron Scientific Consulting Inc.

85

SIGNIFICANCE OF BURN-IN IN ELECTRICAL CONTACTS

Selected Burn-In Methods

- Reciprocating motion of pin in socket while passing current, but without contact disruption possible deleterious effects:
 - generate unwanted mechanical wear on contact surfaces and removal of thin protective electroplates
 - increase a permanent set in receptacle springs
 - possible arcing
 - other effects

© Timron Scientific Consulting Inc.

86

SUMMARY

Major Parameters and Mechanisms Affecting Contact Resistance

- I. SurfaceRoughness: Asperity density and shape can optimize connector function
- 2. Surface Hardness: Hardness determines real contact area
- 3. Interdiffusion: Usually deleterious to contact performance
- 4. Electroplates: To modify surface hardness and provide protection against mechanical wear and corrosion; underplates reduce interdiffusion between electroplates and substrate

© Timron Scientific Consulting Inc.

SUMMARY

Major Parameters and Mechanisms Affecting Contact Resistance

- 5. Surface Insulating Films: Usually deleterious to contact performance since they add to contact resistance; these films may increase susceptibility to fretting corrosion
- 6. A-spot Temperature: Controls interdiffusion processes and other mechanisms such as oxidation and corrosion rates; elevated temperatures are usually deleterious to contact performance. Temperature can be evaluated from the V-T relation

© Timron Scientific Consulting Inc.

88

SUMMARY

Major Parameters and Mechanisms Affecting Contact Resistance

- 7. Signal Frequency: The "skin effect" begins to have a noticeable effect on connection resistance at a frequency of a few MHz
- 8. Small Contacts: Classical contact theory breaks down for a-spot radii smaller than a few hundred nanometers
- Contact Degeneration Mechanisms: Oxidation, corrosion, fretting corrosion, intermetallic growth, differential thermal expansion etc.. eventually limit connector life.

© Timron Scientific Consulting Inc

89